

Product nam	е	ClearView Yellow Aid
Revision date		6-11-15
Product ID:		Yellow Aid
Chemical Name: Synonyms: Chemical Formula CAS Number: Product Use: Supplier:	ı: Oreq Corporation 42306 Remington Temecula, CA 92 951-296-5076	
Emergency Phone	2# (	Chemtrec: 1-800-424-9300
GHS classification Labels and other t	=	Not Classified Not Classified

Com	ponents	CAS No.	Weight %
Sodiu	m Bromide	7647-15-6	98-100

Eye contact	Hold eye open and rinse slowly and gently with water for 15-20 minutes. Remove contact lenses, if present, after the first 5 minutes, then continue rinsing eye. Call a poison control center or doctor for treatment advice.
Skin contact	Take off contaminated clothing. Rinse skin immediately with plenty of water for 15-20 minutes. Call a poison control center or doctor for treatment advice.
Inhalation	Move person to fresh air. If person is not breathing, call 911 or an ambulance, then give artificial respiration, preferably by mouth-to-mouth, if possible. Call a poison control center or doctor for further treatment advice.
Ingestion	Call poison control center, or doctor immediately for treatment advise. Have person sip a glass of water if able to swallow. Do not induce vomiting unless told to do so by the poison control center or doctor. Do not give anything by mouth to an unconscious person.
Most important symptoms and effects, acute or delayed	
Note to physician	None known No specific antidote. Treat symptomatically and supportively. Probable mucosal damage may contraindicate the use of gastric lavage.



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Suitable extinguishing media	Material is not combustible. Use extinguishing media appropriate to surrounding fire conditions.	
Unusual fire and explosion hazards	Will decompose from ca. 800°C releasing poisonous and corrosive fumes of hydrogen bromide and sodium oxide.	
Fire fighting procedure	Cool containers with water spray. In closed stores, provide fire-fighters with self- contained breathing apparatus in positive pressure mode	
Personal precautions	Use dust respirator, rubber gloves and chemical safety goggles.	
Methods for cleaning up	Sweep up, place in a bag and hold for waste disposal or possible re-use Ventilate area and wash spill site after material pickup is complete. Avoid raising dust.	
Environmental precautions	Prevent entry into sewers and watercourses	
Handling	Avoid bodily contact. Keep containers tightly closed.	
Storage	Keep in a well-ventilated place away from incompatible materials (see "materials to avoid").	

### Exposure Limits :

Exposure Limits :		1
Components	ACGIH-TLV Data	OSHA (PEL) Data
Sodium Bromide	Not determined	Not determined
7647-15-16		
Ventilation requirements	Provide adequate ventilation.	
Personal protective equipment: - Respiratory protection - Hand protection - Eye protection - Skin and body protection	In case of significant or accidental dust em Protective gloves Chemical safety goggles Body covering clothes and boots	issions, dust mask should be worn
Hygiene measures	Do not eat, smoke or drink where material is handled, processed or stored. Wash hands carefully before eating or smoking. Safety shower and eye bath should be provided.	
Appearance Melting point/range Boiling point/range Flash point Evaporation rate (ether=1) Flammable/Explosion limits	White, odorless, crystalline solid 755°C 1390°C None Not applicable under standard conditions Not flammable	



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Vapour pressure Vapor density Solubility:	1 mmHg (806°C) Not applicable under standard conditions
<ul> <li>Solubility in water</li> <li>Solubility in other solvents</li> </ul>	94.6 gr/100ml at 25°C ethanol: 95%: 7 g/100g at 25°C methanol: 14.8 g/100g at 25°C
Partition coefficient (n-octanol/water) Auto-ignition temperature Viscosity: Specific gravity Explosive properties Oxidizing properties	Not applicable since this material is almost completely soluble in water. Not applicable Not applicable 3.203 Not explosive Not oxidizing
Reactivity	Reacts explosively with bromine trifluoride .
Stability	Stable under normal conditions The powder product tends to cake under normal storage conditions.
Possibility of hazardous reactions	Not expected to occur
Conditions to avoid	Heating above decomposition temperature
Materials to avoid	Strong acids. Strong oxidants. Heavy metal salts.
Hazardous Decomposition Products	Hydrogen bromide and sodium oxide Bromine fumes
Likely Routes of Exposure	Skin Eye contact Inhalation Ingestion
Acute toxicity: - Rat oral LD50 - Rabbit dermal LD50 - Rat dermal LD50 - Eye irritation (rabbit) - Dermal irritation (rabbit)	4200 mg/kg >2000 mg/kg >2000 mg/kg Slightly irritant Not irritant
Dermal sensitization	Not a sensitizer
Chronic toxicity	Repeated skin contact may cause dermatitis.Repeated oral intake of bromides (>9 mg/kg body weight/day) may affect the central nervous system. Warning symptoms include mental dullness, slurred speech, weakened memory, apathy, anorexia, constipation, drowsiness and loss of sensitivity to touch and pain.
Mutagenicity	Does not induce DNA repair in cultured human epithelioid cells. Not clastogenic in human lymphocytes metaphase analysis.



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	Not mutagenic by the Ames Test
Carcinogenicity	Not classified by IARC / Not included in NTP 13th Report on Carcinogens
Reproductive toxicity	Sodium bromide has been shown to cause embryo-fetal toxicity and malformations in rats at dose levels which also produce maternal toxicity. The No-Observed Effect Level (NOEL) is 100 mg/kg/day, and the Acceptable Daily Intake (ADI) for sodium bromide from food and drinking water in humans is 1 mg/kg/day. Comparable high doses of sodium chloride (table salt) similarly cause malformations, embryo-fetal toxicity, and maternal toxicity in mice.
Teratogenicity	In the oral gavage pre-natal developmental toxicity study in the Rabbit, there were no obvious effects of maternal treatment on the survival, growth or development of the offspring at any of the dosages investigated. The No Observed Effect Level (NOEL) for the developing conceptus was considered to be 250 mg/kg/day.
Aspiration hazard	Not expected to occur
Environmental fate	NaBr is an inorganic salt, which fully dissociates in aquatic environment to bromide and sodium ions. It also undergoes degradation in soil to bromide ion (no further degradation or biodegradation will occur).
Aquatic toxicity : - 96 Hour-LC50, Fish	>1000 mg/l (rainbow trout) >1000 mg/l (bluegill sunfish)
- 48 Hour-EC50, Daphnia m	na >1000 mg/l
Avian toxicity: - Oral LD50, Bobwhite quai - Dietary LC50, Mallard duc - Dietary LC50, Bobwhite qu Toxicity to micro-orga	
Toxicity to micro-orga	NOEC was 1000 mg/l (3 hours)
Persistence and degra Bioaccumulative poter Mobility in soil	
Waste disposal 0	oserve all federal, state and local environmental regulations when disposing of this material.
Packaging disposal	spose of in a safe manner in accordance with local/notional regulations.
DOT	Not regulated
IMDG	Not regulated



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ICAO/IATA	Not regulated	
USA	Reported in the EPA TSCA Inventory This product is registered under FIFRA.	
- Emergency overview in accordance to EPA Master Label	Sodium Bromide-SP (EPA Reg. No. 8622-78) CAUTION Causes eye irritation Harmful if absorbed through skin or swallowed This product is toxic to fish and aquatic organisms.	
	Sodium Bromide-COMP (EPA Reg. No. 8622-45) CAUTION Harmful if absorbed through skin or swallowed This product is toxic to fish and aquatic organisms.	
- SARA 313	Not listed	
Canada	Listed in DSL	
-WHMIS hazard class	D2A very toxic material causing other toxic effects	
EU	Reported in EINECS	
EC No.	231-599-9	
Japan	ENCS no. 1-113 ISHL no. 1-113	
Australia	Listed in AICS	
New Zealand Inventory	Listed in NZIoC	
China inventory	Listed in IECSC	
Korea	Listed in ECL (KE-31368)	
Philippines	Listed in PICCS	

#### DATE OF PREPARATION 6-11-15

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